

 **S1. Particle analysis using ImageJ.** Fluorescence images were converted to 8-bit, and a thresholding of 30–175 was used to allow for better visualization of fluorescent particles. Following this, the "analyze particles" function was used to measure all the particles on each image and generate the results of the areas and Feret diameters of each particle (Scheider et al., 2012). The particles on each image were outlined and labeled. To avoid background noise from the PCTE filters under the 40x magnification, a particle size threshold was used in the "analyze 32 particles" function to only measure and count the particles that had areas larger than 79  $\mu$ m<sup>2</sup> 33 (in a perfect circular shape, diameter  $= 10 \text{ µm}$ ) (Scheider et al., 2012). However, the particle's shape was not necessarily a perfect circle, so – after the data was imported to RStudio, a final 35 selection was made using Feret  $\geq 10$  µm as the final cut-off size reported. As MP fibers do not fluoresce in the same manner as MP fragment particles do, the fibers were measured by length and color in the bright field microscopic images using the "line tool" in ImageJ. These data were also automatically generated for each filter (Klein & Fischer, 2019; Scheider et al., 2012).

## **Equations**.

41 *Relative response factor (RRF)* = 
$$
\frac{Slope\ of\ indicator\ ion\ of\ each\ polymer}{Slope\ of\ calibration\ curve\ of\ cholanic\ acid}
$$
 (S1)

42 *MP mass* = 
$$
\frac{Peak\ area\ of\ polymer's\ indicator\ ion\times mass\ of\ cholanic\ acid}{Peak\ area\ of\ quantifier\ ion\ of\ cholanic\ acid}\times\frac{1}{RRF}
$$
 (S2)

 **S2. Positive Controls.** Microplastic standards were subjected to the same filtration, digestion, 45 and analysis procedures as the atmospheric deposition samples. Both polyethylene (size  $\sim$  500  $\mu$ m) and polyvinylchloride (size ~100  $\mu$ m) particles were added to a 2 L glass bottle containing Type I water and then filtered onto a PCTE membrane. The PCTE membrane containing the 48 positive controls was subjected to the same  $H_2O_2$  digestion process as the samples, as described in Section 2.3. The positive controls were then dyed with Nile Red and examined using fluorescence microscopy (Section 2.4). The microplastic standards were recovered, giving confidence to the validity of the sample preparation and analysis methods.

54 **Table S1.** Pyrolysis-GC/MS parameters.



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- 58 **Table S2.** Indicator ions and retention time and index of standard common consumer plastics
- 59 determined in this study. The retention index is Kováts Retention Index, which was
- 60 calculated using pyrolysis production of PE (alkenes).



 **Table S3**. The parameters resulting from the polymer calibration curves. The low value provided in the range column can be considered the method limit of quantification for each polymer. RRF = relative response factor compared to cholanic acid (internal standard). Note: due to heterogenous ball milling and solvent solubility issues, a calibration curve for Nylon6 67 could not be completed. Thus, the RRF used for Nylon6 (12.7) was derived from Klein  $\&$ Scholz-Böttcher (2017). All Nylon6 data should be considered semi-quantitative.



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- 73 **Table S4**. The background signals detected in the Pyr-GC/MS analysis from control samples
- 74  $(n=3)$ .



77 **Table S5.** The mass and polymers determined for the individual particles analyzed using Pyr-



78 GC/MS. Note: Nylon6 data should be considered semi-quantitative.



81 **Table S6**. The Pearson correlation coefficients between MP deposition rates and



82 meteorological factors. The asterisks  $(*)$  denote a P value < 0.05.

83



**Figure S1**. The 48 h air parcel back trajectories at 100 m calculated using the HYSPLIT

- model available from NOAA for rain events recorded in Auckland over the sampling period.
- Map tiles by Stamen Design, under CC BY 3.0. Data by OpenStreetMap, under ODbL.